



COURSE OUTLINE

The course description is online @ <http://camosun.ca/learn/calendar/current/web/chem.html>

Ω Please note: the College electronically stores this outline for five (5) years only.
It is **strongly recommended** you keep a copy of this outline with your academic records.
You will need this outline for any future application/s for transfer credit/s to other colleges/universities.

1. Instructor Information

Instructor:	Diana Li
Office Hours:	Mon, Wed, & Thu 14:30-15:20, Mon & Wed 16:30-17:20, or by appointment.
Location:	F344C
Phone:	250-370-3444
Email:	lid@camosun.bc.ca
Website:	http://camosun.ca/learn/programs/chem.html

To avoid class interruptions & cancellations, please refrain from wearing fragrance or other strongly scented products to class!

2. Intended Learning Outcomes

(No changes are to be made to these Intended Learning Outcomes as approved by the Education Council of Camosun College.)

Upon completion of this course the student will be able to:

1. Utilize nomenclature rules to name ionic and covalent compounds.
2. Demonstrate an understanding of stoichiometry by balancing chemical equations and performing mathematical calculations involving chemical reactions.
3. Describe the electronic structure of any atom in the periodic table and apply it to explain many of the physical and chemical properties of the elements.
4. Utilize simple bonding theories to explain why elements combine to form the compounds they do and also to explain many of the properties of compounds.
5. Apply knowledge of intermolecular interactions to rationalize many important physical properties of bulk matter in the gas, liquid and solid phases.
6. Use standard chemistry lab equipment, including burets, pipets, Buchner filters, and volumetric glassware in the correct manner.
7. Perform many standard laboratory procedures, such as titrations, preparation of standard solutions, the preparation, isolation, and purification of compounds, as well as use spectrophotometers to make analytical measurements.

3. Required Materials

Texts	"Chemistry, The Central Science: a broad perspective" by Brown et al., 2010—a.k.a. 2nd Australian Custom BLB Edition.
Other	<ul style="list-style-type: none">◆ Chem 120 Lecture Notes Supplement & Appendix by D. Li, Fall 2013 Ed. This supplement goes with both 1st & 2nd Australian editions of BLB although the 2nd Australian edition is preferred. Note: Homework questions selected from both editions are not necessarily identical but equivalent.◆ Chem 120 Lab Manual◆ Darling Molecular Model Kit—highly recommended if you are taking Chem 121 and other 2nd year Chemistry courses (Chem 230, 231, etc.)◆ Safety glasses & lab coat are both mandatory!

4. Course Content and Schedule

Lecture Plan:		
Chapter		Topic (approximate number of lecture hours)
2 nd Ed.[1 st Ed.]		
1-3	[1-3]	Review & Preview of Selected Topics (9)
9	[9]	Gases (3-4)
5	[5]	Electronic Structure of Atoms (7-8)
6	[6]	Periodic Properties of the Elements (3)
7	[7]	Basic Concepts of Chemical Bonding (3-4)
8	[8]	Molecular Geometry & Bonding Theory (3 + lab lecture)
10	[10]	Intermolecular Forces, Liquids & Solids (3)
11	[11]	Solutions (3)
20	[16]	Chemistry of the Environment (3)

5. Basis of Student Assessment (Weighting)

Labs (9 experiments)	22%
Test I (Review/Preview Topics)	13% (Week VI Lab Periods)*
Test II (Ch. 9, 5 & 6 Aussie Ed.)	20% (Week XI Lab Periods)*
Final Exam (comprehensive)	45% (TBA~Week VII, 3 hours in April)

* Test dates to be confirmed during the first two weeks of classes in January.

Notes:

- (1) Student must pass the lab portion of the course to obtain credit for Chem 120. Your lab faculty will go over the lab component of Chem 120 and lab evaluation with you...
- (2) Student is encouraged to attempt both tests. Test score that is not as high as that of the April final exam will be dropped automatically and its weight redistributed to the final exam. However anyone who is caught cheating will receive zero for that test which will not be redistributed. For anyone who misses both tests, your final exam will then be 78% of the course grade!
- (3) Student must write each test as scheduled for his/her section. No one is allowed to write late and there will be no exceptions. Early exam is a privilege and not a right, thus, at full discretion of the instructor.

6. Grading System

(No changes are to be made to this section unless the Approved Course Description has been forwarded through the Education Council of Camosun College for approval.)

Standard Grading System (GPA)

Percentage	Grade	Description	Grade Point Equivalency
90-100	A+		9
85-89	A		8
80-84	A-		7
77-79	B+		6
73-76	B		5
70-72	B-		4
65-69	C+		3
60-64	C		2
50-59	D	Minimum level of achievement for which credit is granted; a course with a "D" grade cannot be used as a prerequisite.	1
0-49	F	Minimum level has not been achieved.	0

Temporary Grades

Temporary grades are assigned for specific circumstances and will convert to a final grade according to the grading scheme being used in the course. See Grading Policy E-1.5 at camosun.ca for information on conversion to final grades, and for additional information on student record and transcript notations.

Temporary Grade	Description
I	<i>Incomplete:</i> A temporary grade assigned when the requirements of a course have not yet been completed due to hardship or extenuating circumstances, such as illness or death in the family.
IP	<i>In progress:</i> A temporary grade assigned for courses that, due to design may require a further enrollment in the same course. No more than two IP grades will be assigned for the same course. (For these courses a final grade will be assigned to either the 3 rd course attempt or at the point of course completion.)
CW	<i>Compulsory Withdrawal:</i> A temporary grade assigned by a Dean when an instructor, after documenting the prescriptive strategies applied and consulting with peers, deems that a student is unsafe to self or others and must be removed from the lab, practicum, worksite, or field placement.

7. Important Dates

Week

VI Family Day: Feb 10 (Mon)—College Closed

Test I in Lab: Feb 11 (Tue) 14:30-17:00 for 120-02 &
Feb 12 (Wed) 10:30-13:00 for 120-01.

Reading Break: Feb 13 & 14 (Thu & Fri).

X Mar 10 (Mon): Last Day to Withdraw or Change to Audit...

XI **Test II in Lab:** Mar. 18 (Tue) 14:30-17:20 for 120-02 &
Mar. 19 (Wed) 10:30-13:20 for 120-01

April 14: First Day of Exams for Winter 2014

8. Recommended Materials or Services to Assist Students to Succeed Throughout the Course

Articles in the Library Reserve Room for Chem 120 & 121

1st & 2nd Australian Editions of B-L-B & Online resource tool: www.pearsoned.com.au/brown.
(Note: Answers to red & black numbered exercises are available 2 ways—email Di or use CD that is packaged with the new text!)

LEARNING SUPPORT AND SERVICES FOR STUDENTS

There are a variety of services available for students to assist them throughout their learning. This information is available in the College calendar, at Student Services, or the College web site at camosun.ca.

STUDENT CONDUCT POLICY

There is a Student Conduct Policy **which includes plagiarism**. It is the student's responsibility to become familiar with the content of this policy. The policy is available in each School Administration Office, at Student Services, and the College web site in the Policy Section.

Students may not use recording devices in the classroom without the prior permission of the instructor. However, the instructor's permission is not required when the use of a recording device is sanctioned by the College's Resource Centre for Students with Disabilities in order to accommodate a student's disability and when the instructor has been provided with an instructor notification letter which specifies the use of a recording device. Recordings made in the classroom are for the student's personal use only, and distribution of recorded material is prohibited.